

Anzahl der Elektronen auf Außenschale.	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18																																													
	1,0 H Wasserstoff 1	2,0 He Helium 2	3,0 Li Lithium 3	4,0 Be Beryllium 4	5,0 B Bor 5	6,0 C Kohlenstoff 6	7,0 N Stickstoff 7	8,0 O Sauerstoff 8	9,0 F Fluor 9	10,0 Ne Neon 10	SC			11,0 Na Natrium 11	12,0 Mg Magnesium 12	13,0 Al Aluminium 13	14,0 Si Silicium 14	15,0 P Phosphor 15	16,0 S Schwefel 16	17,0 Cl Chlor 17	18,0 Ar Argon 18																																										
19,0 K Kalium	20,0 Ca Calcium	21,0 Sc Scandium	22,0 Ti Titan	23,0 V Vanadium	24,0 Cr Chrom	25,0 Mn Mangan	26,0 Fe Eisen	27,0 Co Cobalt	28,0 Ni Nickel	29,0 Cu Kupfer				30,0 Zn Zink	31,0 Ga Gallium	32,0 Ge Germanium	33,0 As Arsen	34,0 Se Selen	35,0 Br Brom	36,0 Kr Krypton	37,0 Rb Rubidium	38,0 Sr Strontium	39,0 Y Yttrium	40,0 Zr Zirkonium	41,0 Nb Niob	42,0 Mo Molybdän	43,0 Tc* Technetium	44,0 Ru Ruthenium	45,0 Rh Rodium	46,0 Pd Palladium	47,0 Ag Silber	48,0 Cd Cadmium	49,0 In Indium	50,0 Sn Zinn	51,0 Sb Antimon	52,0 Te Tellur	53,0 I Jod	54,0 Xe Xenon																									
55,0 Cs Caesium	56,0 Ba Barium	57,0 bis	58,0 Hf Hafnium	59,0 Ta Tantal	60,0 W Wolfram	61,0 Re Rhenium	62,0 Os Osmium	63,0 Ir Iridium	64,0 Pt Platin	65,0 Au Gold	66,0 Hg Quecksilber	67,0 Tl Thallium	68,0 Pb Blei	69,0 Bi Wismuth	70,0 Po* Polonium	71,0 At* Astat	72,0 Rn* Radon	73,0 Fr* Francium	74,0 Ra* Radium	75,0 bis	76,0 Bi Bismut	77,0 Po Polonium	78,0 At Astat	79,0 Rn Radon	80,0 Fr Francium	81,0 Ra Radium	82,0 bis	83,0 Bi Bismut	84,0 Po Polonium	85,0 At Astat	86,0 Rn Radon	87,0 Fr* Francium	88,0 Ra* Radium	89,0 bis	90,0 Th Thorium	91,0 Pa Protactinium	92,0 U Uran	93,0 Np Neptunium	94,0 Pu Plutonium	95,0 Am Americium	96,0 Cm Curium	97,0 Bk Berkelium	98,0 Cf Californium	99,0 Es Einsteinium	100,0 Fm Fermium	101,0 Md Mendelevium	102,0 No Nobelium	103,0 Lr Lawrencium	104,0 Rf Rutherfordium	105,0 Db* Dubnium	106,0 Sg* Seaborgium	107,0 Bh* Bohrium	108,0 Hs* Hassium	109,0 Mt* Meitnerium	110,0 Ds* Darmstadtium	111,0 Rg* Roentgenium	112,0 Cn* Copernicium	113,0 Nh* Nihonium	114,0 Fl* Flerovium	115,0 Mc* Moscovium	116,0 Lv* Livermorium	117,0 Ts* Tenness	118,0 Og* Oganesson
<p><b>Das Periodensystem der Elemente</b></p> <p><a href="https://goo.gl/3Vf9yZ">https://goo.gl/3Vf9yZ</a></p> <p><b>1,0</b> → <b>Atommasse in u</b></p> <p><b>H</b> → <b>Elementsymbol</b></p> <p>Wasserstoff → <b>Name des Elements</b></p> <p><b>1</b> → <b>Ordnungszahl = Zahl der Protonen</b></p>																			Metalle		Lanthanoide		Erdalkalimetalle		Nichtmetalle		Übergangsmetalle		Actinoide		Edelgase																																
* Nur radioaktive Isotope bekannt.																			57 - 71		89 - 103																																										
Anzahl der mit Elektronen besetzten Schalen.																			Metalle		Lanthanoide		Erdalkalimetalle		Nichtmetalle		Übergangsmetalle		Actinoide		Edelgase																																